

# Ab initio structure prediction for molecules and solids

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## **Contents**

#### structure prediction:

- 1) global search on potential surface minima correspond to (meta)stable structures
- 2) search for barriers

#### method:

simulated annealing: start from atoms, perform moves until solid is formed => change atom positions and lattice constant

examples: LiF (bulk; cluster), BN, CaC<sub>2</sub>, GeF<sub>2</sub>

#### Why ab initio?

model potentials good for ionic systems, less for covalent or metallic systems ab initio calculations more generally applicable, also if type of bond unknown

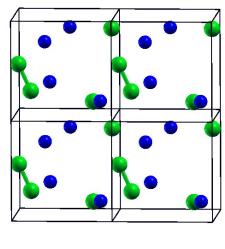
## Method, part I

- I) simulated annealing with "G42" code
  - 0) guess structure -> compute energy (random geometry!)
  - create new structure
     (atoms moved or exchanged, lattice constant changed ...),
     compute new energy
  - 2) accept new structure according to probability  $exp(-\Delta E/kT)$  (Metropolis algorithm)

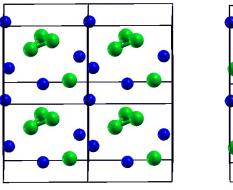
reduce temperature during the run; stop after  $\sim 10^4 \dots 10^6$  steps; followed by quench ( $\sim 10^4$  steps at T=0)

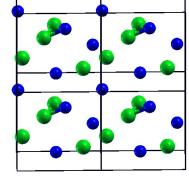
J. C. Schön and M. Jansen, Angew. Chem., Int. Ed. Engl. 35, 1286 (1996)

## **Example**



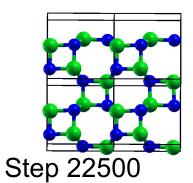
Step 1





Step 2543

Step 2544 (2 atoms exchanged)



## Method, parts II, III

II) local optimisation: start from geometry after quench, optimise with analytical gradients (CRYSTAL code)

analytical gradient with respect to atom positions:

K. Doll, V. R. Saunders, N. M. Harrison, Int. J. Quant. Chem. 82, 1 (2001) and unit cell:

K. Doll, R. Dovesi, R. Orlando, Theor. Chem. Acc. 112, 394 (2004)

III) symmetry analysis with KPLOT (R. Hundt, Bonn)

in total: ~100...1000 runs required (different run=different initialisation of random number generator)

#### **Barrier calculation**

#### Threshold algorithm:

- 1) Start from minimum, allow all moves up to a certain energy (lid)
- 2) perform quench, analyse structure
- => transition to other structure possible

## Ab initio calculations: with CRYSTAL-Code

Idea: use methods from molecular quantum chemistry for periodic systems -> Gaussian type functions used (i.e. local basis set)

1975 Turin: Begin of the code development

1988 first release: CRYSTAL88

further releases:1992, 1995, 1998, 2003, 2006













www.crystal.unito.it

#### CRYSTAL09: some of the new features

phonon dispersion (in CRYSTAL06 for k=0) dielectric properties (static dielectric tensor) empirical correction for dispersion

$$E_{\text{disp}} = -s_6 \sum_{i=1}^{N_{\text{at}}-1} \sum_{j=i+1}^{N_{\text{at}}} \frac{C_6^{ij}}{R_{ij}^6} f_{\text{dmp}}(R_{ij}). \qquad f_{\text{dmp}}(R_{ij}) = \frac{1}{1 + e^{-d(R_{ij}/R_r - 1)}}$$

S. Grimme, J. Comp. Chem. 27, 1787 (2006)

.... various others ....

## Ab initio simulated annealing

Idea: connect code for simulated annealing with *ab initio* code (e.g. CRYSTAL) use *ab initio* energy instead of energy from model potential

#### Problems:

- 1) CPU time
- 2) numerical stability

#### Problem 1: CPU time

e.g. LiF bulk: 4 Li atoms, 4 F atoms, experimental geometry

1 calculation with good parameters: 16 seconds

1 calculation with good parameters, no symmetry: 13 minutes

100000 calculations

for 1 simulated annealing and quench run:

13 minutes\*100000: ~ 2 years

100 runs: 2 centuries

model potential: minutes / hours, with even more steps

## How to speed up the ab initio calculations?

Necessary: fast

reasonably stable (convergence at random geometry required!)

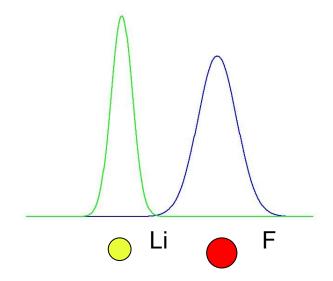
Important: energy does not have to be very accurate

Faster: 1) Gaussian basis set used:

=> increase thresholds for integral selection

e.g. neglect integrals with

Overlap <  $10^{-4}$  (default:  $10^{-6}$ )



- 2) small basis set (no polarisation functions, outermost exponents less diffuse)
- 3) stop SCF cycles earlier (full convergence not necessary)
- 4) fewer k-points ...

## Problem 2: How to achieve stability?

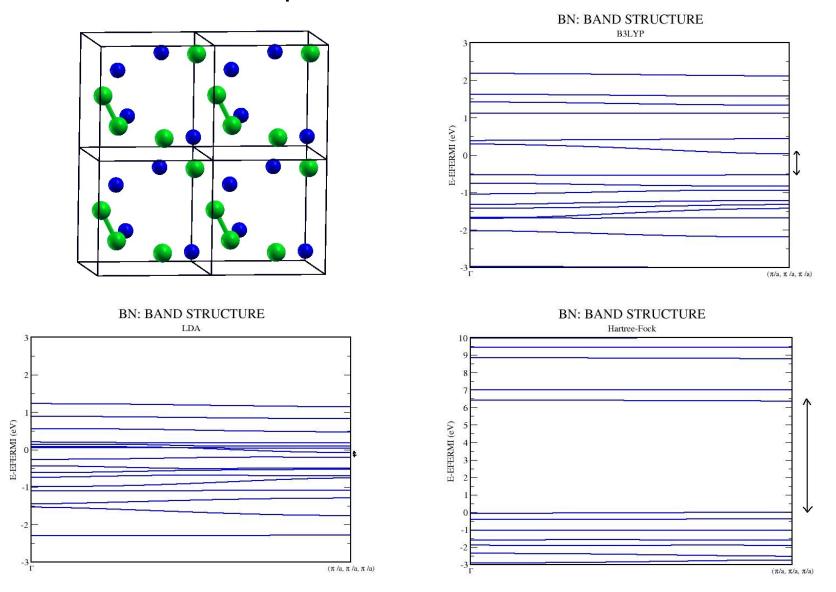
- Difficulties: a) initial geometry is obtained RANDOMLY Kohn-Sham equations have to be solved => expect convergence problems!
  - > 10000 subsequent geometries are generated, calculations performed at all these geometries!
  - b) weak computational parameters, instabilities more likely

## How to achieve stability?

use a method which is not so difficult to converge:

- 1) Hartree-Fock approach quickest: large gap helps to converge
- 2) hybrid functional B3LYP takes much longer than HF
- 3) LDA very difficult to converge (small gap)
- => 1) Hartree-Fock can be used for global optimisation
  - 2) local optimisation is very quick, no problem: compare various functionals, use good parameter values

## Example: BN, initial structure:

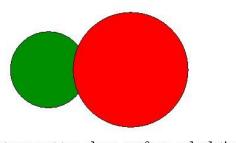


gap: LDA ~0.1 eV; B3LYP ~ 0.5 eV; HF ~6 eV

## How to achieve stability?

atoms too close: discard geometry

discard unrealistic geometries:
minimum distance between atoms required
(use atomic/ionic radii, Mulliken charge)



atoms not too close: perform calculation

if no convergence after maximum number of SCF iterations:

-> discard geometry: assign infinite energy

#### CPU times (one single CPU):

LiF cluster (4 formula units):

Single energy, initial geometry: 0.2 s

(large volume, many integrals discarded)

final geometry: 2 s

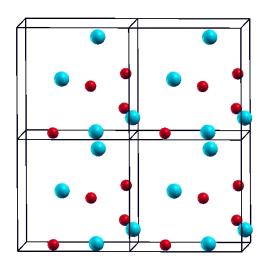
- 1 Simulated annealing run, 45000 steps: a few hours (<10)
- 1 Threshold run for 1 lid, 400000 steps: order of one week

#### CPU times (one single CPU):

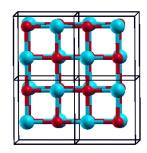
LiF bulk (4 formula units)

single energy, initial geometry: 1 s

(large volume, many integrals discarded)



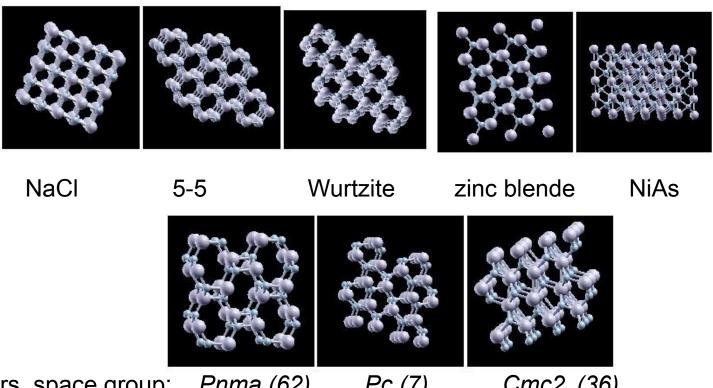
final geometry: 30 s



1 Simulated annealing run, 12500 steps: 3 days

1 Threshold run: ???

#### LiF: proof of principle



Cmc2<sub>1</sub> (36) Pnma (62) *Pc* (7) others, space group:

#### Confirmation of the calculations with model potential

Model potential: J. C. Schön and M. Jansen, Comput. Mater. Sci. 4, 43 (1995) ab initio: K. Doll, J. C. Schön, M. Jansen, Phys. Chem. Chem. Phys. 9, 6128 (2007)

experiment (new LiBr structure: wurtzite):

Y. Liebold Ribeiro, D. Fischer, M. Jansen, Angew. Chemie, 120, 4500 (2008)

## A covalent system: BN bulk

#### Difficulties:

- 1) solution of Schrödinger equation for *random* geometry necessary (much more difficult for covalent system than for ionic system)
- 2) experiment: layered structures (similar to graphite) and "3d" structures (zinc blende,wurtzite)

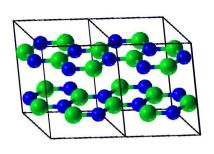
K. Doll, J. C. Schön, M. Jansen, Phys. Rev. B 78, 144110 (2008)

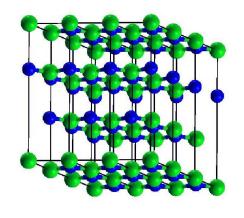
## BN

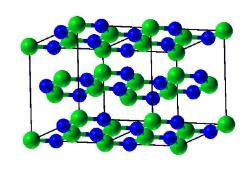
structures found: wurtzite, zinc blende, hexagonal BN ...., **no** roc











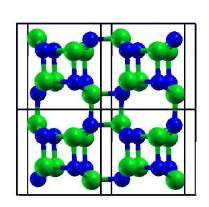
P6<sub>3</sub>/mmc (194)

R3m (160)

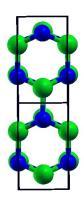
P6m2 (187)

space group P4<sub>2</sub>/mnm:

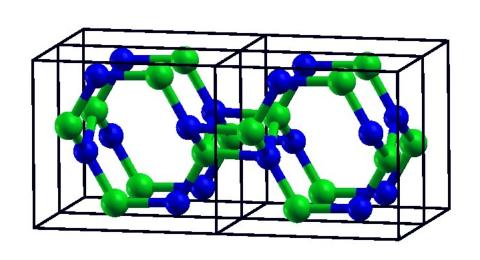
 $\beta$ -BeO (isoelectronic with BN)



different view:



#### SrAl<sub>2</sub> like structure:



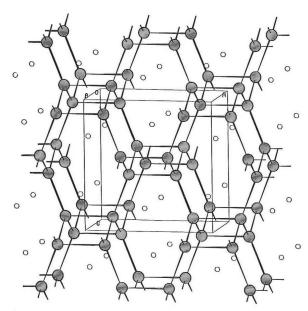


Abb. 2. Perspektivische Ansicht der Struktur der SrAl<sub>2</sub>(ND)-Modifikation. Gestrichelte Kreise  $\triangle$  Al-Atome, kleine Kreise  $\triangle$  SrAtome.

SrAl<sub>2</sub> is isoelectronic: ignore Sr <sup>2+</sup> , then: Al isoelectronic with B

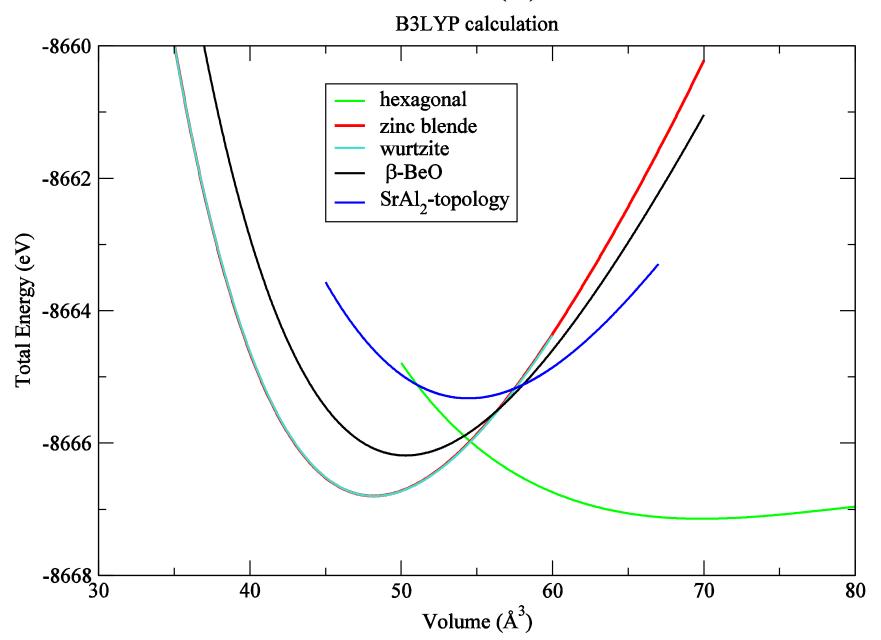
Al 2- isoelectronic with N

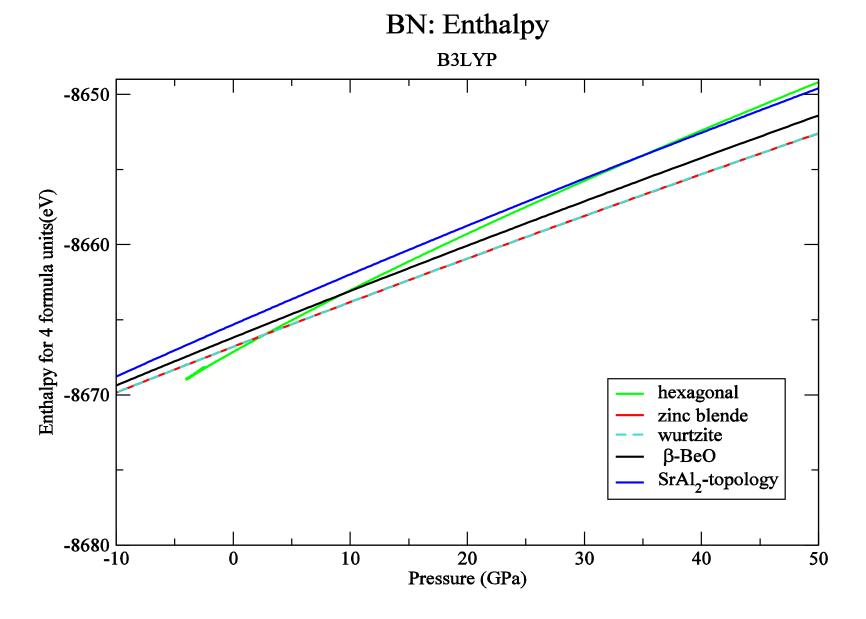
note: BN has space group *Pnma* (62) SrAl<sub>2</sub> has space group *Imma* (74)

G. Cordier, E. Czech, H. Schäfer, Z. Naturforsch. 37 b, 1442 (1982)

identified with TOPOS software (V. A. Blatov, A. P. Shevchenko, V.N. Serezhkin, Samara, Russia)

BN: E(V)

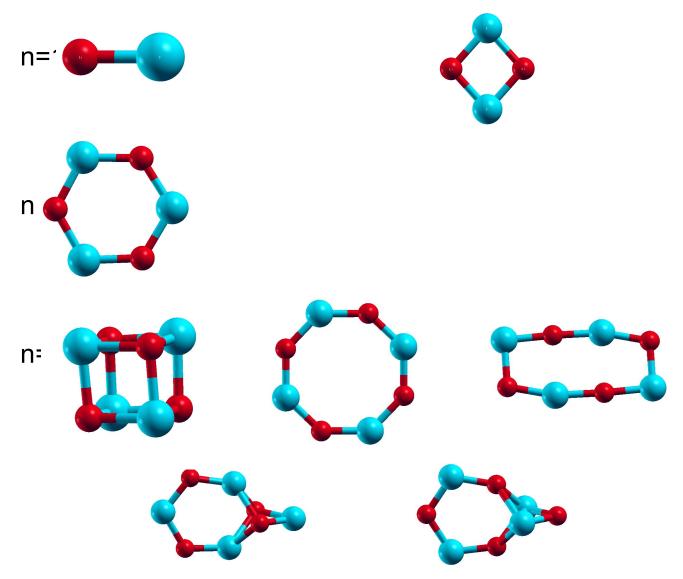




Synthesis of the new  $\beta$  - BeO phase: not feasible by just applying pressure

# (LiF)<sub>n</sub> Clusters

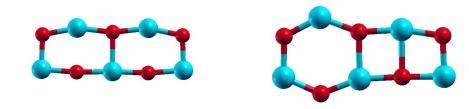


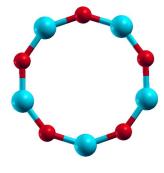


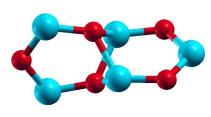
n=5

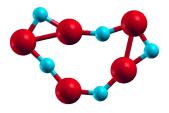


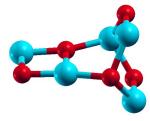


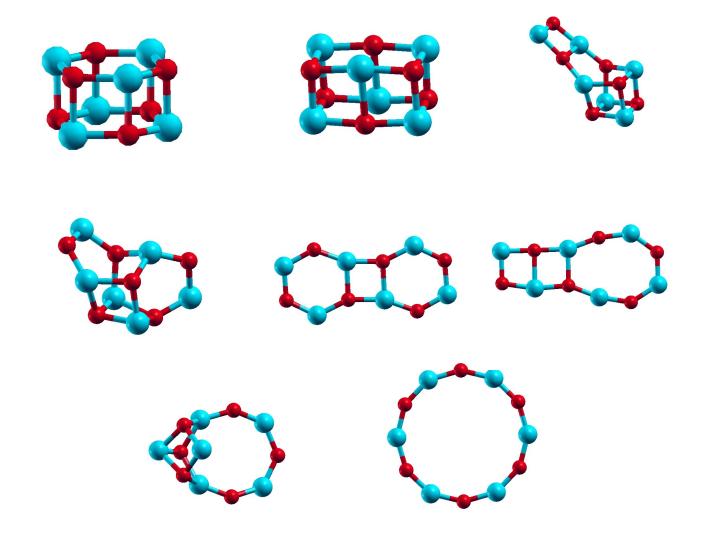


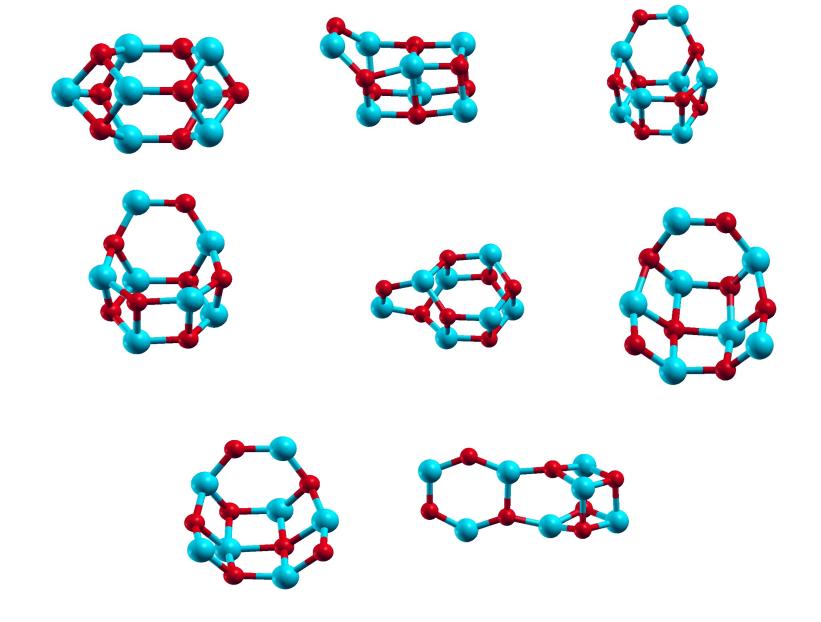


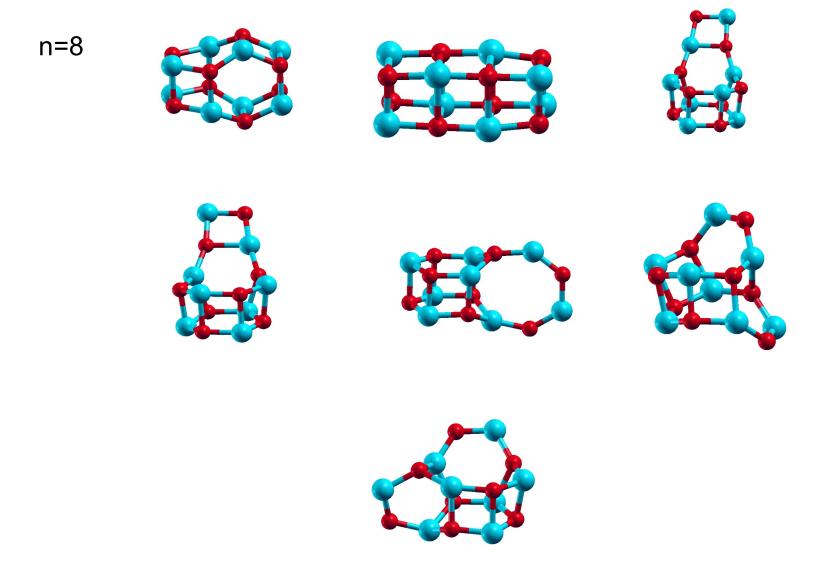










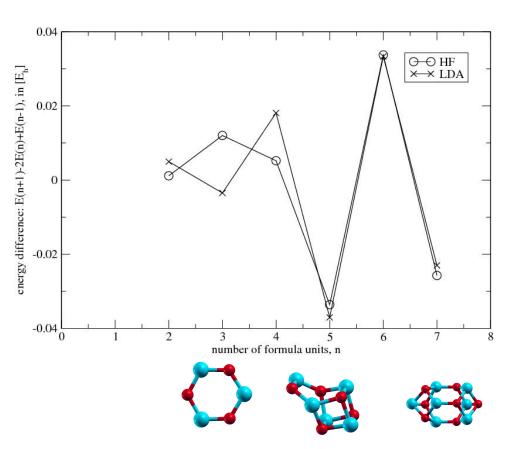


important: the relevant minima are found

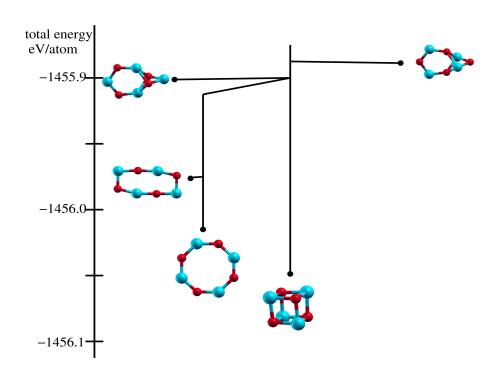
# Clusters (LiF)<sub>2 n</sub> are more stable:







#### Tree graph from threshold run



two lowest lying minima with large barriers → observable?

K. Doll, J. C. Schön, M. Jansen, J. Chem. Phys. 133, 024107 (2010)

#### Conclusion

structure prediction based on simulated annealing and ab initio energies in all the steps feasible

LiF: ab initio results agreed with earlier results based on empirical potentials (proof of principle)

BN: example for a covalent system, new modifications

CaC<sub>2</sub>: mixed covalent/ionic system

GeF<sub>2</sub>: chain like+3d-structures

LiF clusters: barriers can be computed