

# Integration of covalent and non-covalent functionalized SWCNTs in FETs



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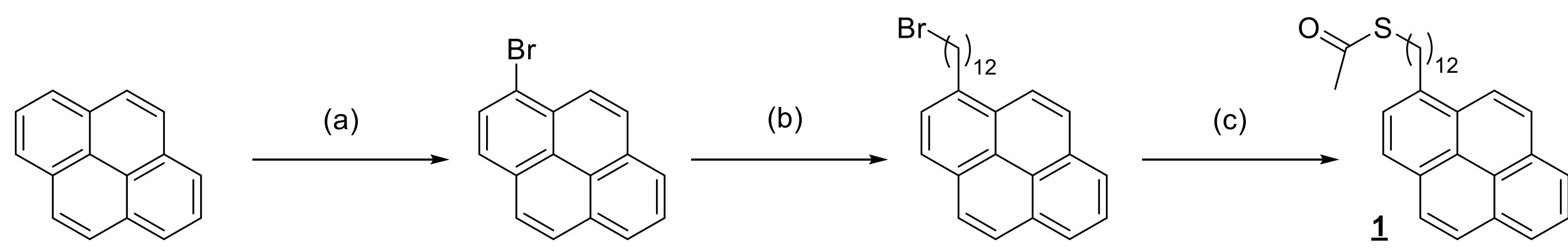


## Motivation

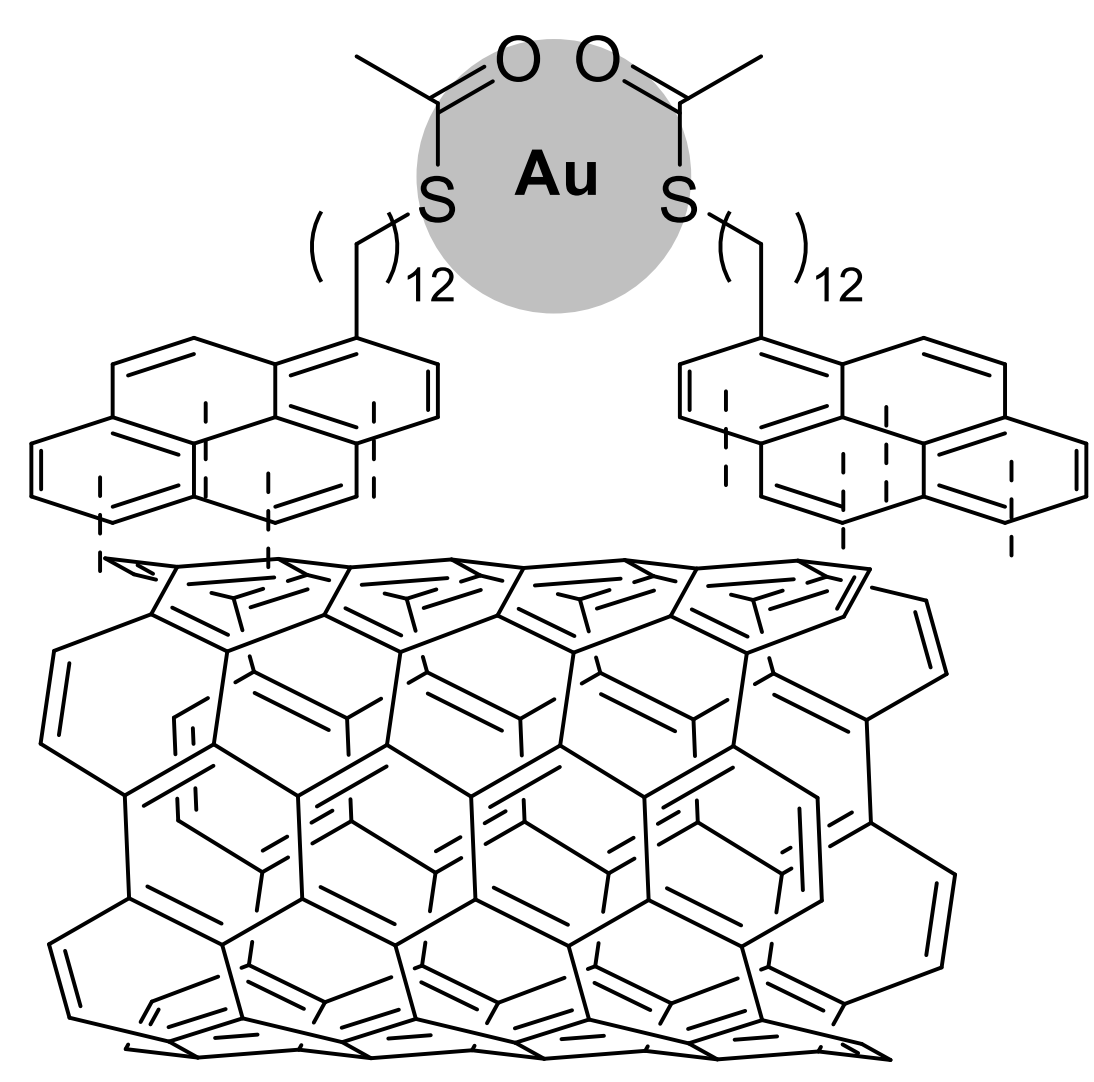
Metal nanoparticles (M-NPs) attached to carbon-based nanostructured materials enable new nanoelectronic solutions for, e.g. energy storage, such as fuel cells or supercapacitors [1] and also for chemical, biochemical [1,2,4] and optical sensors [2,3,4]. A requirement for electronic sensors is the design of a versatile nanoelectronic transducer. In the ideal case, such a component can be functionalized with nanoscopic building blocks in a modular manner that allows selective response and tuning of the sensitivity of the device. Nanoelectronic field-effect transistors (FETs) using individualized single-walled carbon nanotubes (SWCNTs) have been proposed as a FET channel material [5,6].

Recently, we presented a scalable on-chip functionalization approach for SWCNTs between palladium electrodes in the geometry of a FET with preformed Au-NPs based on the Hirsch-functionalization of SWCNTs [5]. This method is wafer-level compatible and comprises two stages of flow chemistry. In a new chemical approach, we propose the deposition of dispersed SWCNTs by using alkylthioate functionalized pyrene which allows selective attachment of M-NPs.

## Non-covalent functionalization

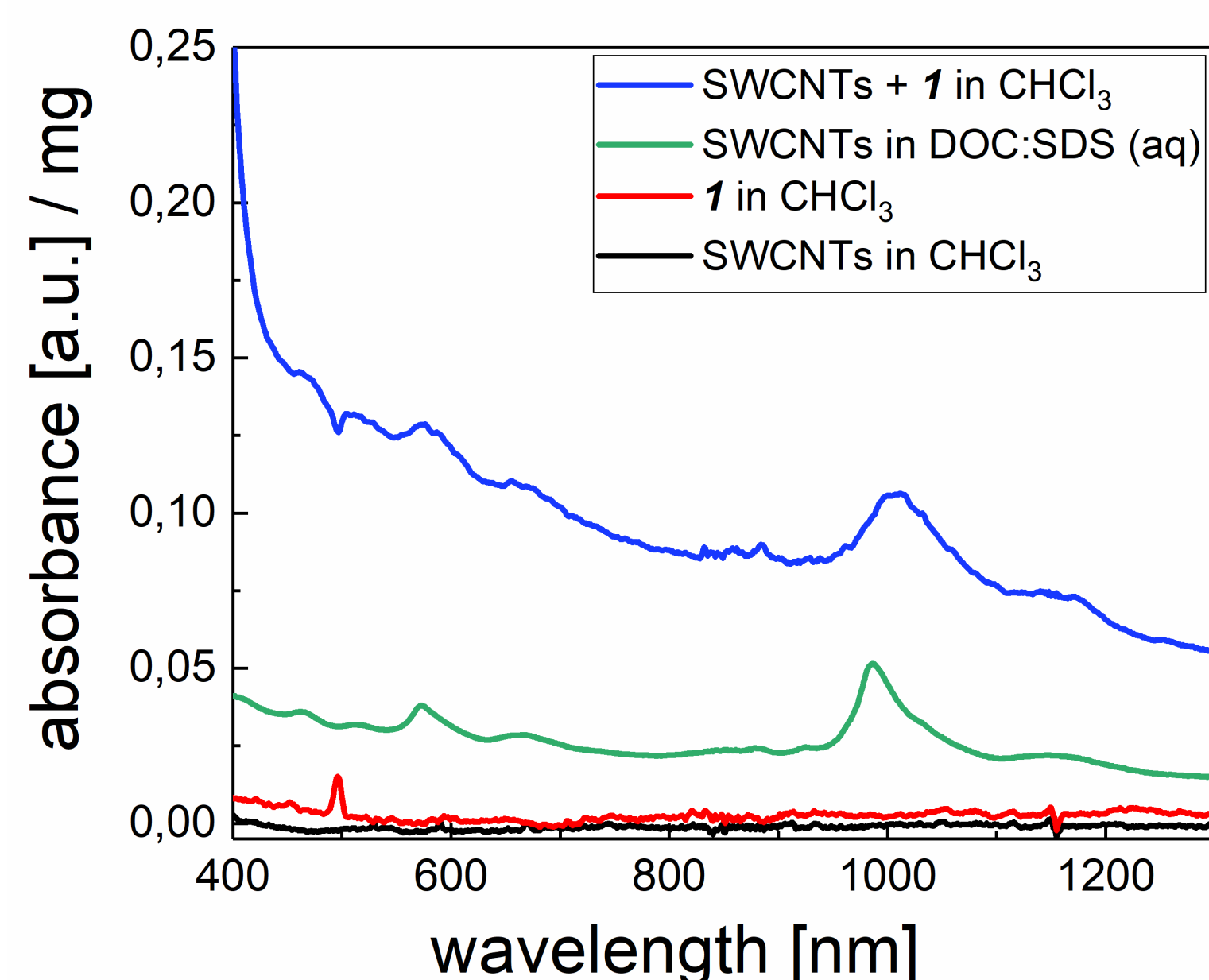


**Scheme 1:** Synthesis protocol for the preparation of **1**.



**Scheme 2:** Schematic representation of a possible linkage between **1** and CNTs by  $\pi$ - $\pi$ -stacking.

- Preparation of SWCNT dispersions (Sigma Aldrich (6,5)-SWCNTs, 80 % purity) in a  $\text{CHCl}_3$ -solution of **1**
- No debundling of SWCNTs by using ultrasonic bath sonication (BS, Fig. 1, blue line)
- Absorption band at 1000 nm for tip sonication (TS, Fig. 1, green line) → Indication of CNT debundling ( $S_{11}$  transition of semiconducting SWCNTs)



**Figure 1:** UV-Vis/NIR spectra of **1** and SWCNTs (Reference system: SWCNTs dispersed with SDS).

## Summary & Outlook

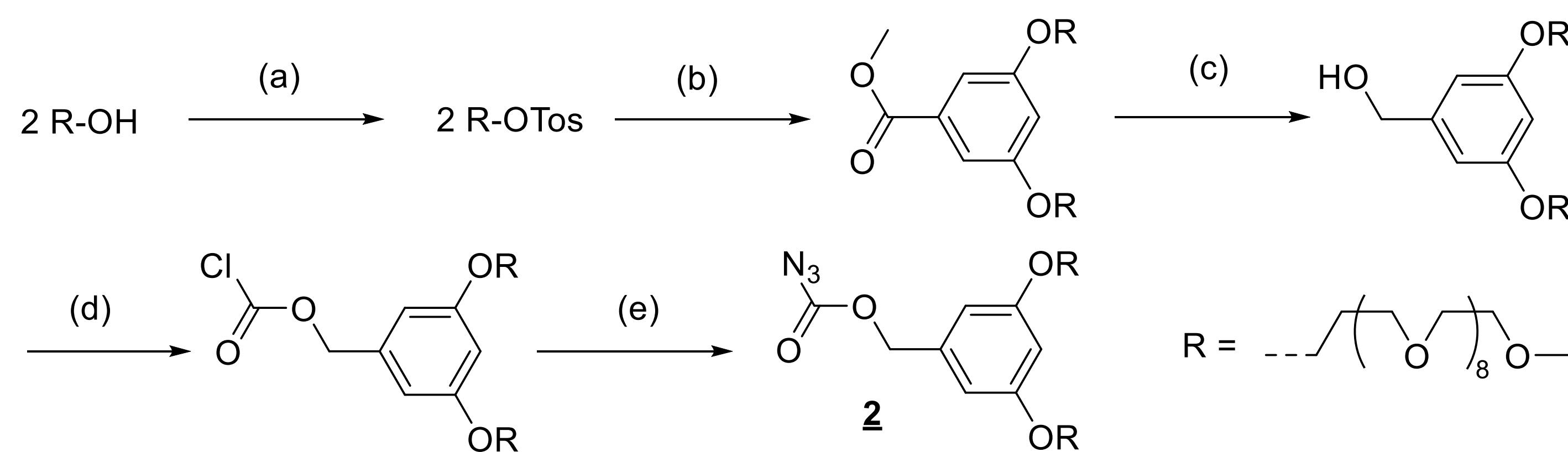
### Summary:

- Debundling of SWCNTs in organic solvents supported by alkylthioate functionalized pyrene through non-covalent interaction**  
→ Access to stable SWCNTs dispersions for integration in FETs
- Au-NP functionalization of SWCNTs by usage of azidobenzyl formiates**  
→ Interaction of Au-NPs with ethylene glycol chains
- Formation of hexagonal gold agglomerates by redoxchemical reactions in water dispersion of Cu-NPs**  
→ Scalable synthesis of hexagonal Au nanostructures

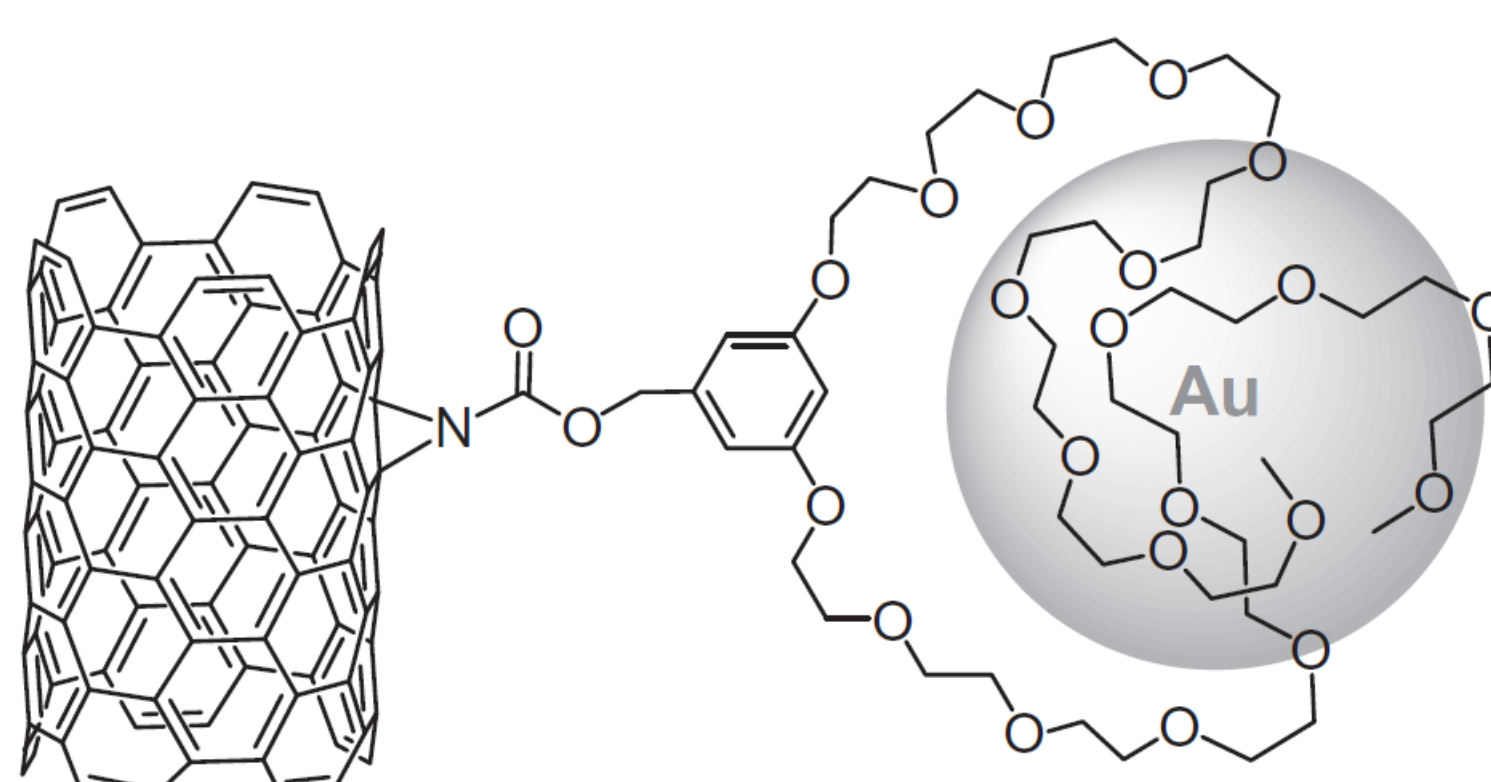
### Outlook:

- Selective attachment of Au-NPs by usage of alkylthioate functionalized pyrene on SWCNTs and integration on wafer-level**  
→ Take advantage of thiophilic behavior of Au-NPs

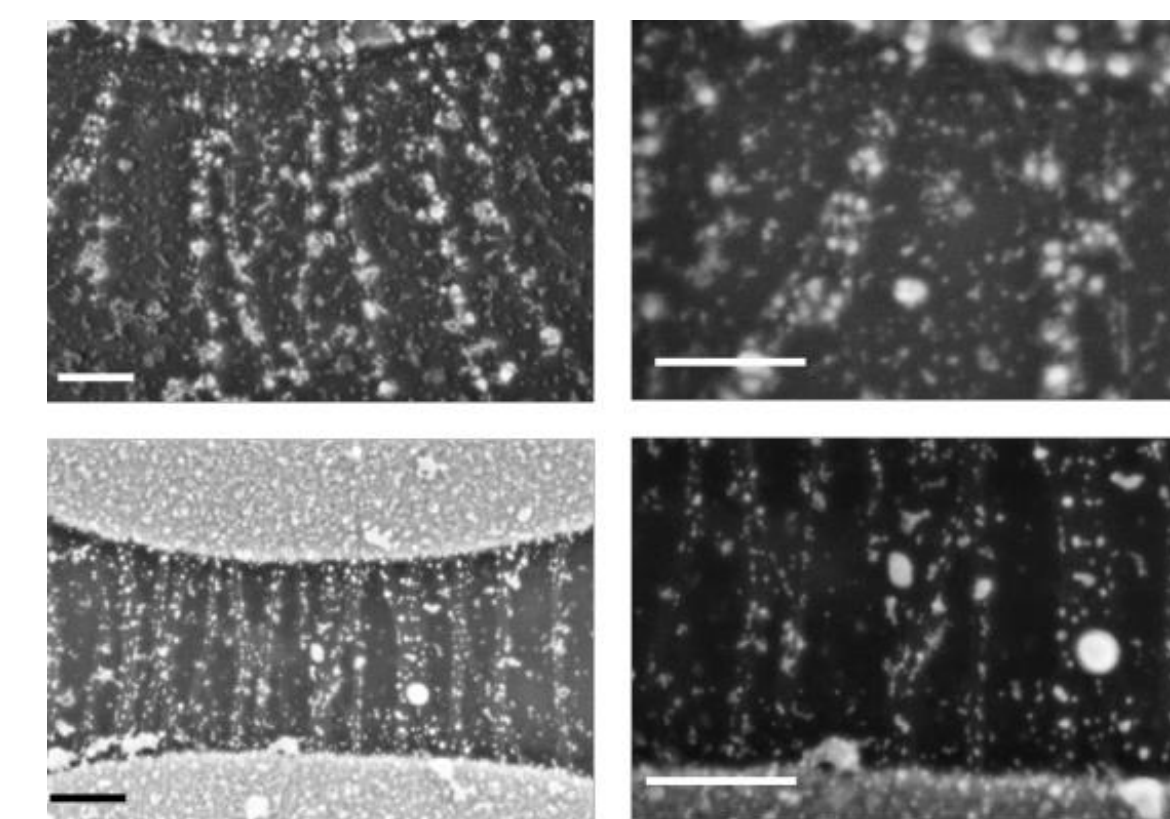
## Covalent functionalization



**Scheme 3:** Synthesis protocol for the preparation of **2** [5,7].



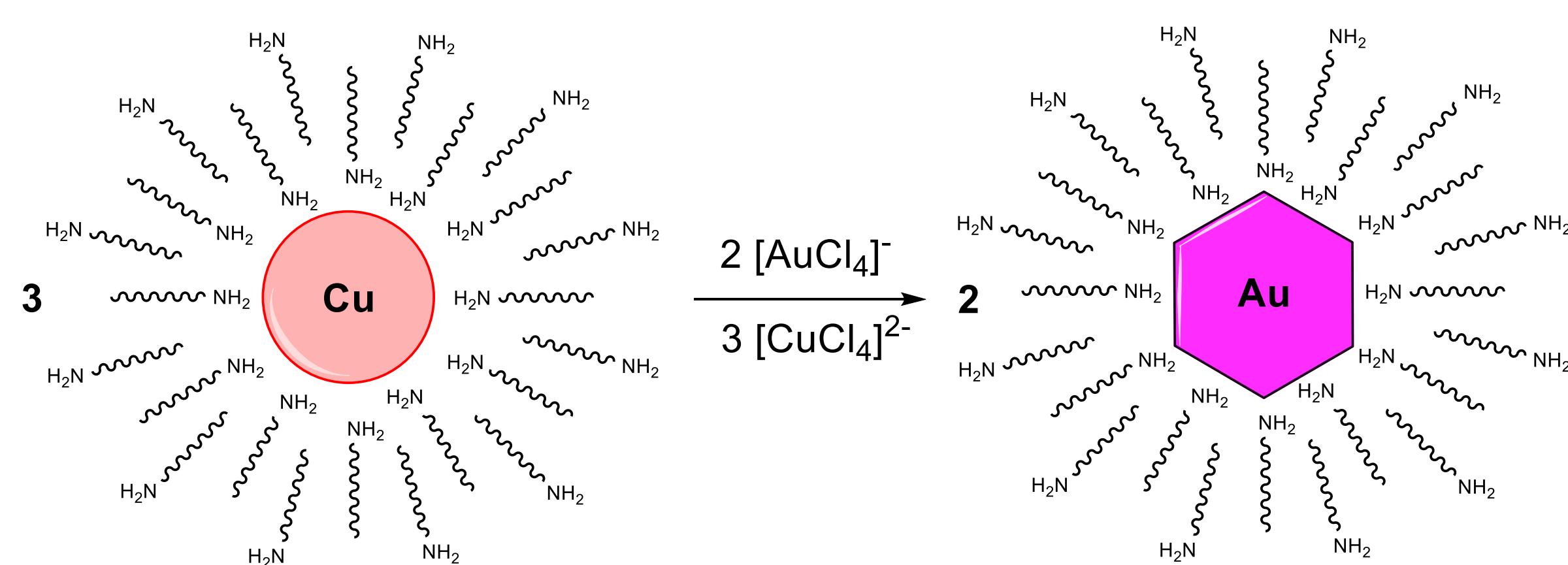
**Scheme 4:** Schematic representation of linkage between **2** and CNTs by [2+1]-cycloaddition [7].



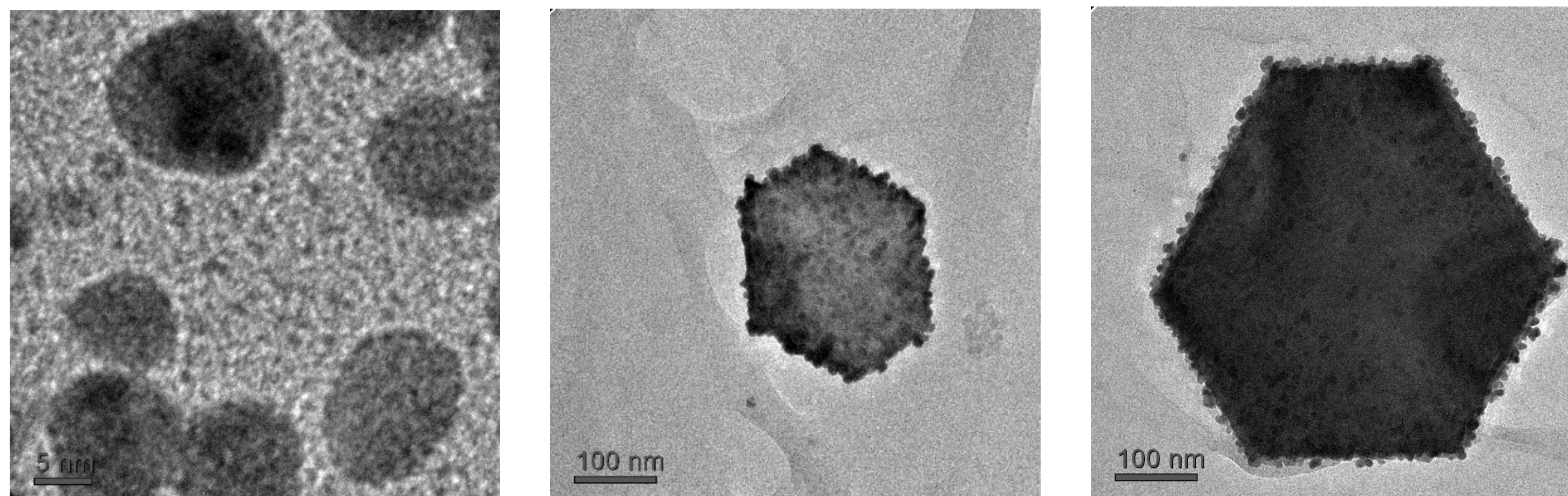
**Figure 2:** REM images (scale bar: 200 nm) of Au-NP decorated CNTs functionalized at 23 °C (top) and 40 °C (bottom) [7].

- No resolution of SWCNTs and ethylene glycol chains by REM (Fig. 2)
- Linear alignment of Au-NPs is observable at 23 °C and 40 °C → Indication of successful CNT functionalization
- Partial agglomeration of Au-NPs

## Synthesis of hexagonal gold nanoparticles



**Scheme 5:** Representation of gold nanoparticle formation in aqueous Cu-NP dispersion.



**Figure 3:** Au-NPs synthesized by redox-chemistry in aqueous Cu-NP dispersions at various  $\text{Au}^{3+}$  concentrations (left to right: 0.05/0.50/5.00  $\text{mmol}\cdot\text{L}^{-1}$ ).

- Formation of spherical Au-NPs at lower  $\text{Au}^{3+}$  concentrations (5–10 nm)
- Hexagonal agglomerates from Au-NPs at higher  $\text{Au}^{3+}$  concentrations (150–500 nm)

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