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Astrochemistry: Recent Successes and Current Challenges  
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## Deuterium Fractionation and Ion-Molecule Reactions at Low Temperatures

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### Abstract

Understanding deuterium fractionation is currently one of the greatest challenges in astrochemistry. In this contribution deuteration experiments of the series  $\text{CH}_n^+$ ,  $n=2-5$ , in a low temperature 22-pole ion trap are used to systematically test a simple chemical rule predicting which molecular ion undergoes deuterium exchange in collisions with HD.  $\text{CH}_4^+$  turns out to be a problem case, where prediction fails. The method of laser induced reaction (LIR) is used to determine the population ratio of the lowest ortho-to-para states of  $\text{H}_2\text{D}^+$  relaxed in collisions with  $\text{H}_2$ . Preliminary results indicate that the ortho-to-para ratio of  $\text{H}_2\text{D}^+$  is substantially reduced in para- $\text{H}_2$ . This points at the important role of nuclear spin in deuterium fractionation, in particular at the destruction of ortho- $\text{H}_2\text{D}^+$  in collisions with ortho- $\text{H}_2$ . More systematic LIR experiments are needed for a chemical model of deuterium fractionation including state-to-state modifications of the species involved.

**Key Words:** astrochemistry — ISM: molecules — molecular processes.

# Deuterium Fractionation and Ion-Molecule Reactions at low Temperatures

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**Abstract.** Understanding deuterium fractionation is currently one of the greatest challenges in astrochemistry. In this contribution deuteration experiments of the series  $\text{CH}_n^+$ ,  $n = 2..5$ , in a low temperature 22-pole ion trap are used to systematically test a simple chemical rule predicting which molecular ion undergoes deuterium exchange in collisions with HD.  $\text{CH}_4^+$  turns out to be a problem case where prediction fails. The method of laser induced reaction (LIR) is used to determine the population ratio of the lowest ortho to para states of  $\text{H}_2\text{D}^+$  relaxed in collisions with  $\text{H}_2$ . Preliminary results indicate that the ortho to para ratio of  $\text{H}_2\text{D}^+$  is substantially reduced in para- $\text{H}_2$ . This points at the important role of nuclear spin in deuterium fractionation. More systematic LIR experiments are needed for a consistent chemical model of deuterium fractionation.

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## 1. Introduction

About 25 deuterated molecules have been found to date in the interstellar medium, Millar (2003). Much attention has been drawn to recent detections of doubly and even triply deuterated species in cold interstellar clouds. Triply deuterated ammonia, Lis *et al.* (2002), and methanol, Parise *et al.* (2004), show an isotopic enhancement (fractionation) which is up to twelve orders of magnitude larger than the expected value based on the cosmic D/H ratio of  $1.5 \times 10^{-5}$ . Recent detection of  $\text{D}_2\text{H}^+$ , Vastel *et al.* (2004), stimulated new chemical modelling calculations for very cold dense clouds which find that even  $\text{D}_3^+$  can be the dominant molecular species, Walmsley *et al.* (2004), Flower *et al.* (2004).

The main source of deuterium in molecular clouds is HD. Three primary molecular ions,  $\text{H}_3^+$ ,  $\text{CH}_3^+$  and  $\text{C}_2\text{H}_2^+$  are known to transfer the deuterium in collisions with HD. These reactions are slightly exothermic by a few hundred Kelvin due to differences in zero-point vibrational energies. Since typical cloud temperatures are much smaller than these exoergicities, isotopic enhancement of the singly deuterated ions can easily reach factors up to  $10^4$ . Reactions in subsequent collisions of these ions lead to the deuteration of other molecules like  $\text{HCO}^+$ .

Due to the Arrhenius type behavior of the back reactions (hydrogenation) the D/H ratio strongly depends on the temperature of the particular environment. Also the fractional electron abundance plays an important role for isotopic fractionation as dissociative recombination competes with deuteration. For the same reason the presence of atoms and molecules which react with the primary ions also reduce the isotopic enhancement. These species include almost all C, N and O bearing molecules. In contrast, freeze out of molecules on grains in very cold environments removes these species from the gas

phase. As a consequence isotopic enrichment is found to be increased. Many of these dependencies are included in current chemical models for fractionation.

Another level of complication arises from collision systems with identical nuclei. This is for example the case in all three key initial deuteration reactions. Besides conservation of energy, momentum, angular momentum and parity, conservation of the total nuclear spin imposes further restrictions on the outcome of such reactions, Quack (1977). Because of this, even more importantly, inelastic collisions with the most dominant  $\text{H}_2$  molecule can influence the D/H ratio significantly. Thorough descriptions of appropriate theoretical methods to include nuclear spin effects as well as experiments to investigate the role of nuclear spin have been given by Gerlich *et al.* (2002a), Gerlich & Schlemmer (2002b), and Gerlich (2004).

Current laboratory challenges on understanding deuterium fractionation include (i) the search for more primary deuteration reactions, (ii) testing of theoretical predictions for possible deuteration reactions, and (iii) the role of nuclear spin conservation on the level of isotopic fractionation. More systematic studies on possible deuteration reactions of small hydrocarbons,  $\text{CH}_n^+$ , ( $n=3..5$ ), Asvany *et al.* (2004a), and of  $\text{C}_3\text{H}_n^+$ , ( $n=1..3$ ) have been carried out recently, Savic & Gerlich (2005), Savic *et al.* (2005). In this contribution the series  $\text{CH}_n^+ + \text{HD}$  will be used to compare the results to theoretical predictions worked out by Henchman *et al.* (1988) and modified by Maluendes *et al.* (1992).

In order to judge on the role of nuclear spin conservation state-to-state reactive and inelastic rate coefficients are needed. Experimentally this is very difficult. However, as described below, a method has been developed in recent years, Schlemmer *et al.* (1999), Schlemmer *et al.* (2002), Mikosch *et al.* (2004), Schlemmer *et al.* (2005), Asvany *et al.* (2005a), Asvany *et al.* (2005b), to infer ro-vibrational spectra of a cloud of molecular ions in a trap experiment. Based on a few assumptions the rotational population of the ion cloud can be derived. First results on  $\text{H}_2\text{D}^+$  using the free electron laser FELIX are presented.

## 2. Experimental

To measure the rate coefficients of low-temperature ion-molecule reactions a 22-pole ion trap has been used. This apparatus has been developed and thoroughly described by Gerlich (1992) and recent applications to isotopic fractionation are well documented, Gerlich *et al.* (2002a), Gerlich & Schlemmer (2002b), Asvany *et al.* (2004b), Savic & Gerlich (2005), Savic *et al.* (2005).

In brief, ions are produced in a storage ion source by electron bombardment, mass filtered in a quadrupole analyzer and then injected into the 22-pole ion trap which is held at cryogenic temperatures,  $T > 10$  K. On entrance, the ion temperature is adapted to the trap temperature by a short helium pulse. During the storage time of several seconds the ions can react with a neutral gas, here mainly HD, of known constant but adjustable number density. After the trapping period, the trap content is extracted, mass analyzed and the parent and product ions are counted with high efficiency. From this procedure, the rate coefficients of the involved processes are inferred by repeating the mentioned cycle of *trapping/reaction/detection* for all possible reaction products and different trapping times.

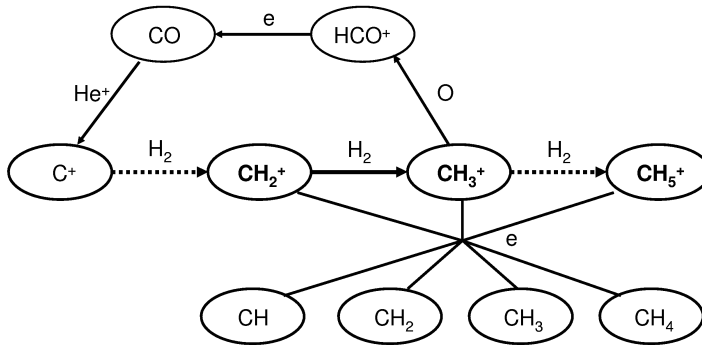
An extension of this technique is the method of laser induced reactions (LIR), in which an endothermic ion-molecule reaction (e.g.  $\text{H}_2\text{D}^+ + \text{H}_2 \rightarrow \text{H}_3^+ + \text{HD}$  as presented below) is enhanced by resonantly exciting the trapped parent ions with a laser. This offers not only the possibility to record a spectrum by counting the product ions as a function of the laser wavelength, but LIR can also yield information on state-selected rate coefficients,

lifetimes of the excited states and the population of the pumped states, Schlemmer *et al.* (2002), Schlemmer *et al.* (2005).

### 3. Results and Discussion

#### 3.1. Kinetic Studies

Fig. 1 shows the network of small hydrocarbon ions as proposed by Black & Dalgarno (1977) for diffuse interstellar cloud conditions. Similar schemes have been found also for cold dense clouds. In this scheme carbon is cycled between the small hydrocarbon ions,  $\text{CH}_n^+$ ,  $n=0..3$ , and CO. Thus these ions play an important role in the molecule formation of interstellar clouds. Starting with  $\text{C}^+$ , hydrogen is either added via radiative association (adding  $\text{H}_2$ , dashed line) or in an abstraction reaction (adding H, solid line). Growth stops at  $\text{CH}_5^+$ , since  $\text{CH}_n^+$  with  $n > 5$ , does not form a strongly bound molecule. Instead, dissociative recombination of this molecule is a major source for interstellar methane.



**Figure 1.** Formation routes for small hydrocarbon molecules in interstellar clouds as described by Black & Dalgarno (1977).

Based on such scheme it is not obvious why only  $\text{CH}_3^+$  should be a source for deuteration. Therefore more systematic studies of the deuteration of the  $\text{CH}_n^+$ ,  $n=2..5$ , ions have been conducted in recent years, Asvany *et al.* (2004a). The results of the kinetic measurements at a trap temperature of 15 K are summarized in Table 1. Various outcomes are possible in the collision of  $\text{CH}_n^+ + \text{HD}$ .  $\text{CH}_2^+$  undergoes hydrogen or deuterium abstraction to form  $\text{CH}_3^+$  or  $\text{CH}_2\text{D}^+$ . This process is so fast, that any deuteration, forming  $\text{CHD}^+$ , can hardly be detected, especially since the abstraction reaction product,  $\text{CH}_3^+$ , has the same mass as the deuteration product,  $\text{CHD}^+$ . Comparison of the measured total rate coefficient for the abstraction reaction to the limiting collisional rate coefficient,  $k_L$ , reveals that this process is happening practically in every collision. In fact the experimental value is about 20 % larger than the predicted value. One possible reason for this discrepancy lies in the long range attraction of HD.

Since abstraction is so fast for  $\text{CH}_2^+$  also in collisions with  $\text{H}_2$ , the less frequent collisions with HD hardly yield a substantial fraction of  $\text{CH}_2\text{D}^+$ . Because of this reason  $\text{CH}_2^+$  does not play a role in isotopic fractionation. Instead only those ions are relevant for isotopic fractionation in collisions with HD which react very inefficiently with  $\text{H}_2$ . This is the case for  $\text{CH}_3^+$  which forms  $\text{CH}_5^+$  only very slowly via radiative association and for  $\text{CH}_5^+$  which is effectively the end of the reaction chain. Therefore at least these two ions are

**Table 1.** Rate coefficients, branching ratios and reaction efficiencies ( $k/k_L$ ) for the deuteration reactions of  $\text{CH}_n^+ + \text{HD}$ ,  $n=2..5$ , at  $T = 15$  K.

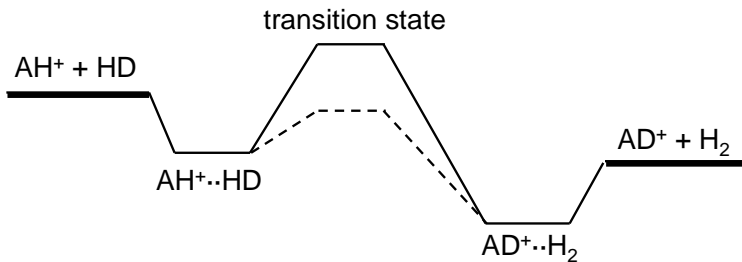
reaction	branching ratio	$k$ [ $10^{-9} \text{cm}^3 \text{s}^{-1}$ ]	$k/k_L$
$\text{CH}_2^+ + \text{HD}$	$\rightarrow \text{CH}_2\text{D}^+ + \text{H}$ $\rightarrow \text{CH}_3^+ + \text{D}$	0.8 0.2	1.6 1.21
$\text{CH}_3^+ + \text{HD} (+\text{He})$	$\rightarrow \text{CH}_2\text{D}^+ + \text{H}_2$		1.65
$\text{CH}_3^+ + \text{HD} (+n\text{-H}_2)$	$\rightarrow \text{CH}_2\text{D}^+ + \text{H}_2$		1.5
$\text{CH}_3^+ + \text{HD} (+p\text{-H}_2)$	$\rightarrow \text{CH}_2\text{D}^+ + \text{H}_2$		0.4
$\text{CH}_2\text{D}^+ + \text{HD} (+\text{He})$	$\rightarrow \text{CD}_2\text{H}^+ + \text{H}_2$		1.59
$\text{CD}_2\text{H}^+ + \text{HD} (+\text{He})$	$\rightarrow \text{CD}_3^+ + \text{H}_2$		1.5
$\text{CH}_4^+ + \text{HD}$	$\rightarrow \text{CH}_3\text{D}^+ + \text{H}_2$ $\rightarrow \text{CH}_5^+ + \text{D}$ $\rightarrow \text{CH}_4\text{D}^+ + \text{H}$	<0.002 0.68 0.32	0.45 0.34
$\text{CH}_5^+ + \text{HD}$	$\rightarrow \text{CH}_4\text{D}^+ + \text{H}_2$		$< 5 \cdot 10^{-9}$

reasonable candidates to contribute to isotopic fractionation.  $\text{CH}_4^+$  is not shown in the network, see Fig. 1 and should be considered separately.

For  $\text{CH}_3^+$  sequential deuteration has been measured and rate coefficients for all individual steps have been determined at a trap temperature of 15 K, see Table 1. As in the case of  $\text{CH}_2^+$  the rate coefficients are significantly larger than the capture rate coefficient. Moreover, each step of deuteration happens slightly faster. Besides these peculiarities it turns out that deuteration of  $\text{CH}_3^+$  is happening practically in every collision. It is quite surprising, that this process can be so fast, because simple statistics would predict a 40 % chance to find the D-atom in HD and a 60 % chance to find it in  $\text{CH}_3^+$  after the collision. The deviation from this rather simple counting argument shows that the potential energy surface of the collision system has to be taken into account to understand the large deuteration efficiency.

As is described in detail in Asvany *et al.* (2004a) the results for collisions with pure HD have been compared to experiments where the dominant collision partner is  $\text{H}_2$ . In case of the pure HD,  $\text{CH}_3^+$  has been cooled in the trap to the ambient temperature via collisions with He. The value derived from this experiment is very similar to the value when  $\text{CH}_3^+$  is cooled in n- $\text{H}_2$  which contains 1/4 p- $\text{H}_2$  and 3/4 o- $\text{H}_2$  as well as the terrestrial abundance of  $[\text{HD}]/[\text{H}_2] = 3 \times 10^{-4}$ . However, when conducting the experiment in p- $\text{H}_2$  with the natural abundance of HD, Gerlich & Schlemmer (2002b), the deuteration reaction is found to proceed about four times slower than compared to the other experiments, see Table 1. The presence of the various collision partners changes the relaxation of the ion cloud. This result is linked to the influence that nuclear spin conservation imposes on the rotational state distribution when most collisions are with p- $\text{H}_2$  rather than with o- $\text{H}_2$  or HD. This finding will be discussed below in more detail.

The collision of  $\text{CH}_4^+$  with HD leads primarily to hydrogen abstraction, see Table 1, which has been described in a previous publication, Asvany *et al.* (2004b). The total rate coefficient for abstraction at 15 K, however, amounts to about 30 % of the collision limit. Interestingly, the experimental data show no indication of deuteration in this case. Although the detection of deuterated species has been limited by some spurious isotopomers coinciding with the mass of the products an upper limit for the rate coefficient of deuteration could be derived. According to this result, see Table 1, more than 1000 collisions are necessary to exchange a hydrogen for a deuterium in  $\text{CH}_4^+$  in colli-



**Figure 2.** Energetics of an exoergic deuterium exchange reaction with barrier (solid) without a barrier (dashed).

sions with HD. Another interesting result concerns the branching ratio in the abstraction reaction. The less exoergic product  $\text{CH}_5^+$  is formed twice as often as compared to the energetically more favorable  $\text{CH}_4\text{D}^+$  product. Moreover, the abstraction reaction shows a negative temperature dependence, meaning, that abstraction becomes more favorable at low temperatures. These results indicate that the dynamics of the  $\text{CH}_4^+ + \text{HD}$  collision system does not follow a simple downhill reaction without barrier.

Finally the collision system  $\text{CH}_5^+ + \text{HD}$  has been studied. Like for  $\text{CH}_4^+$  no indication for the exchange of a deuterium for a hydrogen atom has been found. In fact, product detection in this case is more sensitive than for the  $\text{CH}_4^+$  measurements. Therefore the upper limit of deuteration has been determined to be much lower,  $k \geq 5 \times 10^{-18} \text{cm}^3/\text{s}$ , see Table 1. As pointed out before,  $\text{CH}_5^+$  is the end of the hydrogenation route in interstellar clouds, see Fig. 1. The four valence electrons of the carbon atom and the four electrons of the hydrogen atoms are forming a closed shell molecule. Therefore abstraction of D or H does not occur in collisions with HD. Since radiative association is a very slow process for this system,  $k_r = 2 \times 10^{-16} \text{cm}^3/\text{s}$ , dissociative recombination dominates under interstellar conditions, leading to the various neutral small hydrocarbons as depicted in Fig. 1.

In summary,  $\text{CH}_n^+$  exhibits fast deuteration in collisions with HD only for  $n=3$ . All other molecular ions show a negligible chance for the exchange of a deuterium atom. Therefore no additional primary deuteration reaction has been found in these experiments. However, due to the systematic studies presented the results can be compared to predictions or at least simple chemical rules. Such rules have been first formulated by Henschman *et al.* (1988). Here reactions are characterized distinguishing reactants with filled and unfilled valence shells. According to this work, no barriers inhibit deuterium exchange for exoergic reactions if at least one of the reactants has an unfilled valence shell. In contrast, barriers can inhibit the exchange if both reactants have filled valence shells. This simple rule is illustrated in Fig. 2. Here the minimum energy path for an exoergic deuterium exchange,  $\text{AH}^+ + \text{HD} \rightarrow \text{AD}^+ + \text{H}_2$ , is depicted. At the low temperatures of interstellar clouds the two reactants form an entrance complex  $\text{AH}^+ \cdot \text{HD}$ , where the two partners remain as distinct molecules. The H-D exchange needs a closer approach of the two reactants and occurs via a transition state. In case of closed shell species this transition state is substantially higher in energy than the entrance channel. Therefore the exchange reaction is hindered. However, in case of one open shell reactant the complex formed is much stronger bound and the transition state is expected to be below the entrance channel, such that the exchange happens within the collision complex. Due to the exoergic nature of the reaction the collision complex dissociates preferentially into  $\text{AD}^+$  and  $\text{H}_2$  based on statistical arguments counting the number of energetically accessible states. This simple classification is able to explain why  $\text{H}_3^+$ ,  $\text{CH}_3^+$  and  $\text{C}_2\text{H}_2^+$  do undergo H-D exchange in collisions with HD while  $\text{HCO}^+$ ,  $\text{N}_2\text{H}^+$  and  $\text{NH}_4^+$ , for example, do not.

The fact that the "filled valence shell" ions form only weakly bound complexes is related to the fact that electronic reorganization is hindered since empty  $AH^+$  orbitals are energetically inaccessible due to a large HOMO-LUMO ("highest occupied molecular orbital", "lowest unoccupied molecular orbital") gap. Therefore the terms "filled" and "unfilled valence shells" have been replaced by Maluendes *et al.* (1992) for the chemically more precise terms of large and small HOMO-LUMO gap, respectively.

HD has a very large HOMO-LUMO gap, therefore only the small hydrocarbon ions have to be considered in the present case. In the  $CH_n^+$ ,  $n = 2..5$ , series only  $CH_5^+$  has a fully occupied valence shell with a very large HOMO-LUMO gap. The experimental result that only one in more than  $10^8$  collisions lead to H-D exchange is in very good agreement with the prediction. For  $CH_2^+$  the abstraction reaction is so fast that deuterium exchange could not be detected. Therefore no comparison with the simple prediction can be made. H-D exchange is very fast for  $CH_3^+$  which is in perfect agreement with the prediction. In fact, the  $CH_4D^+$  complex which is formed is a strongly bound molecule in its ground state. Therefore the collision complex, the transition state and the complex in the exit channel are more or less the same entity. Apparently, the various H-atoms and the D-atom swap places in this complex very efficiently without substantial energetic hindrance. This process is known as *scrambling*. Thus the picture drawn in Fig. 2 describes this case very well. Even the fact that deuteration is happening in almost 100 % of the cases rather than only in three out of five cases as predicted by the statistical arguments given above has been explained by Maluendes *et al.* (1992). Upon forming the entrance channel complex the system can reach the transition state with a certain probability,  $x$ . In all other cases, probability  $1-x$ , it re-dissociates back into the reactants. From the transition state the complex can fall back into the entrance channel again or react. From the entrance channel it can again form the transition state etc. In effect the chance of forming the deuterated product reaches almost unity.

In contrast, as an open valence shell ion  $CH_4^+$  should undergo rapid H-D exchange, but does not, as shown above. Only little is known about the electronic structure of this molecule, Jacox (1994). Therefore the HOMO-LUMO gap can not be quantified. This hampers a more quantitative comparison. Despite these limitations, as will be discussed in the following, deuteration is hindered even without a substantial barrier. The exoergic abstraction reaction in a  $CH_4^+ + HD$  collision proceeds faster by at least one order of magnitude at 10 K compared to 300 K. Such a behavior is associated with an extended lifetime of the collision complex in the entrance channel. This longer lifetime enhances substantially the chance to overcome the transition state and reaching the products. Even a small barrier can be surpassed via tunneling. Nevertheless, if the reaction is hindered by a barrier, higher temperatures will increase the reaction probability as well. Thus a reaction over a barrier (barrier height  $E_a$ ) will follow the strongly positive temperature dependence of an Arrhenius type behavior,  $\exp(-E_a/kT)$ , in the high temperature limit. In summary, the temperature dependence can be negative (collision complex) for low temperatures but will be positive (barrier) for high temperatures. For the  $CH_4^+ + H_2$  system the negative temperature dependence prevails up to more than 1000 K based on the trap and selected ion flow drift tube (SIFDT) experiments, Asvany *et al.* (2004b), Federer *et al.* (1985). Such a behavior would imply a very high barrier which makes it difficult for the reaction to occur at low temperatures at all. This is in contradiction to the fact that abstraction proceeds in about 1/3 of the cases. In fact, abstraction of  $CH_4^+ + H_2$  occurs about 100 times faster than in the isoelectronic  $NH_3^+ + H_2$  system which shows a turnover to a positive temperature dependence already at 150 K. Therefore it is concluded that not a barrier is hindering the abstraction reaction of  $CH_4^+ + H_2$ . Rather the transition state seems to form a dynamical bottleneck which is hard to surpass. Note,

that for the abstraction an atomic species has to be liberated to form  $\text{CH}_5^+ + \text{D}$  or  $\text{CH}_4\text{D}^+ + \text{H}$ , indicating a very intimate chemical interaction along this path. Apparently, this path can be followed without a substantial barrier. The products of a hypothetical reverse reaction, e.g.,  $\text{CH}_4\text{D}^+ + \text{H} \rightarrow \text{CH}_3\text{D}^+ + \text{H}_2$  are energetically accessible when starting from a low temperature  $\text{CH}_4^+ + \text{HD}$  collision. Therefore deuterium exchange is obviously not hindered by a barrier. Nevertheless it does not occur. This finding is in sharp contrast to the simple model made by Henschman *et al.* (1988), where the hinderance of the H-D exchange shall be associated with a barrier in the transition state.

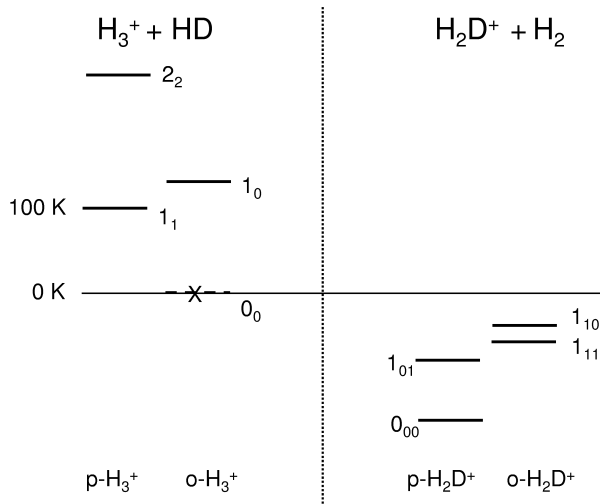
### 3.2. Laser Induced Reactions

In the series of small hydrocarbon ions only  $\text{CH}_3^+$  undergoes rapid H-D exchange. It has been found in the kinetics studies above that the speed of this exchange depends on the nature of inelastic collision partners, p- $\text{H}_2$ , o- $\text{H}_2$  or He, see Table 1. It is the aim of this section to (i) understand the possible reasons for this as well as the consequences on isotopic fractionation and (ii) to describe an experiment which allows to derive rotational populations of a low temperature ion cloud.

Inelastic collisions with  $\text{H}_2$  are much more frequent in space than collisions with HD. Thus  $\text{CH}_3^+$  relaxes to the lowest rotational states accessible. Due to the Pauli exclusion principle antisymmetric/symmetric rotational states of this Fermion are linked to the symmetric/antisymmetric nuclear spin state. As a consequence the two nuclear spin configurations can be treated as separate molecules when dealing with inelastic collisions which do not change the nuclear spin state of  $\text{CH}_3^+$ . In contrast, there is an interchange between the two configurations when nuclear spin changing collisions are permitted. The fast deuteration shows that scrambling of the protons (and the deuteron) is a common process in the  $\text{CH}_4\text{D}^+$  collision complex. This complex decomposes into different nuclear spin configurations, thus allowing for nuclear spin changes. It has been worked out by Bates (1991) for the  $\text{CH}_5^+$  system that conservation of the total nuclear spin is one restriction for the possible outcome of such collisions. As a consequence, relaxation of  $\text{CH}_3^+$  in  $\text{H}_2$  leads to different rotational states of  $\text{CH}_3^+$ , depending on the ortho to para ratio of the  $\text{H}_2$  collision partner. Apparently this must be the reason why deuteration is slower when  $\text{CH}_3^+$  is relaxed in p- $\text{H}_2$  rather than in He or n- $\text{H}_2$ .

Determination of the rotational population of such an ion cloud would yield direct insight into the role of the nuclear spin configurations. Such an experiment has been described recently by Mikosch *et al.* (2004). In this work a 22-pole trap experiment has been devised to determine the rotational population of a  $\text{H}_3^+$  ensemble at  $T = 55$  K by means of action spectroscopy. For this purpose  $\text{H}_3^+$  is excited with more than two quanta of vibration in the  $\nu_2$ -mode using a diode laser. Chemical probing in collisions with Ar is used to detect the excitation. The number of  $\text{ArH}^+$  products is a direct measure for the number of  $\text{H}_3^+$  found in the rotational state prior to excitation. Three rotational lines have been detected. From the relative intensities of the absorption lines the rotational temperature and from the Doppler width the kinetic temperatures have been derived. As a result  $\text{H}_3^+$  is found to be thermalized.

It has been pointed out by Gerlich *et al.* (2002a) that the relative population of the lowest ortho state of  $\text{H}_2\text{D}^+$  plays a key role for the efficiency of the reverse reaction of deuteration of  $\text{H}_3^+$ , see Fig. 3. In effect the back reaction becomes exoergic in collisions of o- $\text{H}_2\text{D}^+$  with o- $\text{H}_2$  since it adds some 175 K of rotational energy. Therefore it is most interesting to determine the rotational distribution of  $\text{H}_2\text{D}^+$ . This has been achieved in the present work by ro-vibrational excitation of  $\text{H}_2\text{D}^+$  using the free electron laser FELIX, in the range of the fundamental  $\nu_2$  and  $\nu_3$  vibrational modes. In this way some 3500 K of energy are added enhancing the back reaction with n- $\text{H}_2$  substantially. De-



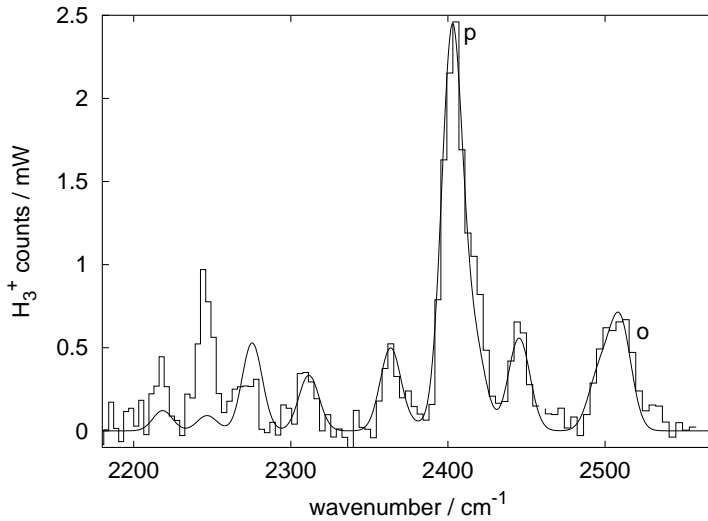
**Figure 3.** State specific energetics of the  $H_3^+ + HD \leftrightarrow H_2D^+ + H_2$  reaction. Only  $J = 0$  states of HD and  $H_2$  are considered. For collisions with,  $J = 1$ ,  $o-H_2$ , 175 K extra energy has to be added.

tecting the number of  $H_3^+$  products as a function of wavelength the spectrum of  $H_2D^+$  shown in Fig. 4 has been recorded. From the intensities of these lines relative rotational populations of the vibrational ground state  $H_2D^+$  could be derived. For the analysis recently calculated Einstein A coefficients of the corresponding transitions have been used, Ramanlal & Tennyson (2004), from which the corresponding Einstein B coefficients have been derived. In addition, it has been assumed that all excited ro-vibrational states undergo hydrogenation with the same efficiency and that the rate of fluorescence, A, is much smaller than the collision rate at  $[H_2] = 7 \times 10^{10} \text{cm}^{-3}$ . At an internal temperature between 20 and 30 K, the population ratio of the lowest ortho and para state of  $H_2D^+$  is about  $\sim 3/4$ . Preliminary experiments using  $p-H_2$  show much smaller values ( $\sim 1/5$ ). These results are in agreement with the general trend that  $p-H_2$  leads to a more efficient relaxation than  $n-H_2$ . This is an important result hinting at the crucial role of nuclear spin in low temperature collisions. However, much higher populations of the lowest ortho state of  $H_2D^+$  would be expected in both cases based on previous studies, Gerlich *et al.* (2002a).

Before final conclusions can be drawn several more systematic studies are necessary. The validity of the assumptions made for the analysis have to be checked in future experiments, e.g., by determining the dependence of the signal on the  $H_2$  number density, thus comparing fluorescence, dominating at *low* densities, with collisions, dominating at *high* densities. Moreover, it has to be taken into account that FELIX is a pulsed laser system with micro-pulses of several picosecond duration only. Therefore saturation effects can influence the experimental outcome. In order to overcome such a limitation, experiments with continuous laser excitation are underway. Also a narrower bandwidth excitation allows for the determination of the kinetic temperature of the ion cloud.

#### 4. Conclusions

Deuteration of  $CH_4^+$  in collisions with HD shows that the simple classification of "un-filled" and "filled valence shells" even in the more sophisticated version of Maluendes *et al.* (1992) can not be used as a general guidance to distinguish reactants which do



**Figure 4.** LIR spectrum of  $\text{H}_2\text{D}^+$  in collisions with  $n\text{-H}_2$ . The simulation (solid line) represents contributions from the  $\nu_2$  and  $\nu_3$  vibrational modes. Transitions of the lowest ortho (o) and para (p) state used for the analysis are marked.

or do not undergo H-D exchange. In contrast, this example demonstrates that not even the knowledge of particular points along the minimum energy path is sufficient. Due to the dynamical restrictions occurring along the reaction path dynamical calculations are necessary to predict the fate of the reactants.

*Scrambling* of H and D is an important way for efficient H-D exchange as seen for  $\text{CH}_3^+ + \text{HD}$  because atoms can swap places without the need of breaking and forming the very strong bonds of the reactants. This process is linked to the chemical equivalence of the atoms involved. Conservation of nuclear spin in collisions with identical nuclei has to be considered. Therefore state-to-state rate coefficients are needed. One experimental route towards this aim is the determination of rotational state populations. Only several hundred  $\text{H}_2\text{D}^+$  are needed to detect a ro-vibrational LIR-spectrum. It is possible to determine the rotational population from such spectra based on several assumptions. A first example spectrum has been analyzed. However, more systematic LIR studies are necessary. Once reliable rotational state populations have been derived at low temperatures more extensive modelling of the  $\text{H}_3^+$  and  $\text{CH}_3^+$  systems can be used to derive state-to-state rate coefficients for inelastic and reactive collisions. With the advent of SOFIA and the Herschel satellite several observations detecting transitions between the lowest rotational states of  $\text{H}_2\text{D}^+$  and  $\text{D}_2\text{H}^+$  are planned. It will be interesting to see how experiment, astrophysical modelling and observation compare.

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## Discussion

KROES:  $\text{CH}_4^+ + \text{HD}$  produces mostly  $\text{CH}_5^+$  through a dynamic mechanism (H further away from centre of mass of HD). For which other systems do you expect this mechanism to be important as well?

SCHLEMMER: The effect of producing more hydrogenated than deuterated species has been predicted by David Clary's group for the collision of  $\text{O}^+ + \text{HD}$ . However, no experiments have been carried out so far. Based on the arguments given in their publication one could expect this effect in many systems exhibiting hydrogen abstraction.